## **Short Communication**

# Nucleophilic Substitution Reactions of Azabenzanthrone Derivatives†

## **SUMMARY**

Reaction of 1-cyano-2-substituted amino-3-azabenzanthrone with p-chloroaniline in the presence of potassium hydroxide in dimethyl sulphoxide gave a mixture of the 4- and 6-p-chloroanilino derivatives. Replacement of the cyano group by the hydroxyl group was observed when the reaction was carried out in the absence of the aromatic amine. Reaction of a 3,9-diazaperylene derivative with p-chloroaniline gave rise to a monoanilino derivative.

## 1. INTRODUCTION

In a recent communication<sup>1</sup> the arylamination of 3-bromo- and 3-nitrobenzanthrones in dimethyl sulphoxide in the presence of potassium hydroxide (DMSO/KOH) has been described. The present paper deals with the nucleophilic reaction of 3-azabenzanthrones with arylamines under similar conditions.

## 2. RESULTS

Reaction of 1-cyano-2-morpholino-3-azabenzanthrone (Ia) with p-chloroaniline (PCA) in DMSO/KOH gave rise to two products which were separated and identified as the 4- and 6-substituted compounds (IIa and IIIa). The structures were established by the difference in the

† Abstracted in part from Ph.D. thesis of D. R. Tatke, University of Bombay, 1976.

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Dyes and Pigments 0143-7208/86/\$03-50 © Elsevier Applied Science Publishers Ltd, England, 1986. Printed in Great Britain

bathochromic shift in alkaline medium (130 nm for 4-substituted and 70 nm for 6-substituted compound). The structure of the 6-substituted product was also confirmed by its synthesis from 1-amino-4-chloro-anthraquinone (IV) cyanoacetyl morpholide and POCl<sub>3</sub> and subjecting the product V to nucleophilic displacement. Similarly, reaction of 1-cyano-2-piperidino-3-azabenzanthrone (Ib) with PCA in DMSO/KOH also gave a mixture of 4- and 6-substituted 3-azabenzanthrones (IIb and IIIb).

In a similar fashion, 8-benzamido-1-cyano-2-morpholino-3-azabenz-anthrone (VIa) was reacted with PCA in DMSO/KOH and again two products (VIIa and VIIIa) could be isolated. The structures were established on the basis of the greater polarity of the 4-substituted derivative (VIIa) as indicated during the chromatographic separation. This was similar to the behaviour of IIa as compared with IIIa. The distinction between the two structures on the basis of the shift in  $\lambda_{\text{max}}$ 

under alkaline conditions was not clearcut since the compound VIIa had twin peaks at 515 and 570 nm. On the other hand, the 6-benzamido derivative (IXa) did not undergo any reaction under these conditions. This can be attributed to the greater acidity of the N—H function of the 6-benzamido group as compared with the 8-benzamido group. This results in the ionisation of the 6-benzamido function and delocalisation of the anionic charge with the carbonyl function of the 3-azabenzanthrone which renders the nucleus inert to the nucleophilic attack.

Reaction of 1-cyano-2-morpholino-3-azabenzanthrone (Ia) with KOH in DMSO without any added nucleophile led to the displacement of the cyano group. This was indicated by the formation of the phenolic product which analysed for the structure (Xa) and showed the molecular ion peak  $M^+$  at m/z 332.

Similar reaction of 1-cyano-2-piperidino-3-azabenzanthrone in DMSO/KOH gave the phenolic product (Xb) with displacement of the cyano group.

Finally, reaction of 1,7-dicyano-2,8-dimorpholino-3,9-diazaperylene (XIa) with PCA in DMSO/KOH gave a violet product which was found to be the mono-substituted product (XIIa) as indicated by elemental analysis. The position of substitution was not determined.

Some of the compounds described in this paper were applied to polyester but they did not possess good dyeing and fastness properties.

## 3. EXPERIMENTAL PROCEDURE

All melting points are uncorrected. The visible spectra were recorded on a Beckmann DK-2A spectrophotometer. The mass spectra were recorded on a Varian CH-7 instrument.

The various 1-cyano-2-substituted amino-3-azabenzanthrones were prepared by the reported procedure.<sup>2</sup>

## 3.1. General procedure for the arylamination of 3-azabenzanthrones

A mixture of powdered KOH (2·0 g) and PCA (0·015 mol) in DMSO (15 ml) was stirred for 15 min. The 3-azabenzanthrone derivative (0·003 mol) was added to this and the mixture stirred at room temperature overnight. The reaction mixture was poured into ice-cold water, acidified with conc. HCl and the product which separated was filtered. The solid was washed with Na<sub>2</sub>CO<sub>3</sub> solution, then water, and dried. The yield, m.p., method of purification, molecular formula and absorption data of the various aminated products are given in Table 1.

## 3.2. Preparation of 1-hydroxy-2-substituted amino-3-azabenzanthrones (Xa and Xb)

The 1-cyano-2-substituted amino-3-azabenzanthrone (Ia,b) (1g) was added to powdered KOH (2g) in DMSO (15 ml) with stirring and the reaction mixture stirred overnight. The mixture was poured into ice-cold water, acidified with HCl (1:1) and the product which separated was

TABLE 1
Physical Data of the Various Arylamino-3-Azabenzanthrone Derivatives

Shift
in
\( \lambda^{\max} \)
(nm)

130

9/

9

1

Com- pound	Substituent	Yield (%)	M.p.	Purification <sup>a</sup> eluent	Mol. <sup>b</sup> formula	λ <sub>max</sub> (nm) (in DMF)	log E	$\lambda_{max}$ $(nm)$ $(in DMF$ $2\% aq$ . NaOH)
I-Chloro-2: IIa	-Chloro-2-morpholino IIa 4-(p-chloroanilino)	35	270	C <sub>6</sub> H <sub>6</sub> -EtOAc	C <sub>27</sub> H <sub>19</sub> ClN <sub>4</sub> O <sub>2</sub>	536	3.93	999
IIIa VIIa	6-(p-chloroanilino) 8-benzamido-4- (n-chloroanilino)	32 33	250 266	C,H, C,H, C,H,-EtOAc	C34H24CIN5O2	536 570 515	3.97	612 625
VIIIa XIIa	8-benzamido-6- (p-chloroanilno) X-(p-chloroanilno) diazaperylene	28	300	C,H, C,H,-EtOAc (9:1)	C <sub>34</sub> H <sub>26</sub> CIN <sub>7</sub> O <sub>2</sub>	590 595	4.36	-
<i>I-Cyano-2-piperidino</i> IIb 4( <i>p-</i> ch IIIb 6( <i>p-</i> ch	piperidino 4(p-chloroanilino) 6-(p-chloroanilino)	33	249	C <sub>6</sub> H <sub>6</sub> C <sub>6</sub> H <sub>6</sub> –EtOAc (95:5)	C34H24CIN5O2	545 540	4.10	615
" Chromatc	<sup>a</sup> Chromatographed over neutral alumina using chlorobenzene (minimum) as solvent <sup>b</sup> Satisfactory elemental analysis data were obtained.	nina using were obtai	chlorobe ned.	azene (minimum)	ıs solvent.			

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chromatographed over neutral alumina using chlorobenzene as solvent and eluting with benzene—ethyl acetate (1:1). 1-Hydroxy-2-morpholino-3-azabenzanthrone (**Xa**) was obtained in 54% yield, m.p. 232–234°C (EtOAc). (Found: C, 71·9; H, 5·0; N, 8·2;  $C_{20}H_{16}N_2O_3$  requires C, 72·3; H, 4·8; N, 8·4%.) 1-Hydroxy-2-piperidino-3-azabenzanthrone (**Xb**) was obtained in 48% yield, m.p. 201–202°C. (Found: N, 8·3;  $C_{21}H_{18}N_2O_2$  requires N, 8·5%.)

## 3.3. Synthesis of 6-(p-chloroanilino)-1-cyano-2-morpholino-3-azabenz-anthrone (IIIa)

A mixture of 6-chloro-1-cyano-2-morpholino-3-azabenzanthrone (IIIa)<sup>3</sup> (0·0015 mol) [obtained by the Vilsmeier reaction on 1-amino-4-chloro-anthraquinone using cyanoacetylmorpholine and POCl<sub>3</sub> m.p. 258 °C (EtOAc)], PCA (0·0018 mol) and anhydrous NaOAc (0·5 g) was refluxed in ethanediol (10 ml) for 30 min. The product, obtained in 68 % yield on adding the reaction mixture to water, was crystallised from benzene, m.p. 251–252 °C. It was found to be identical (m.p., TLC and IR) to the product obtained from the first fraction obtained during the chromatographic separation of the amination product of Ia.

## **ACKNOWLEDGEMENTS**

We are thankful to Professor M. R. Padhye for recording the visible spectra and to Dr K. Nagarajan of Ciba-Geigy Research Centre, Bombay, for the mass spectra.

#### REFERENCES

- 1. T. Vaidyanathan and S. Seshadri, Dyes and Pigments, 5, 431 (1984).
- 2. D. R. Tatke and S. Seshadri, *Indian J. Chem.*, 22B, 1197 (1983).
- 3. K. V. Srinivasan, M.Sc. (Tech.) Thesis, University of Bombay, 1973.

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(Received 24 June, 1985)

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